

CHBE320 LECTURE IV MATHEMATICAL MODELING OF CHEMICAL PROCESS

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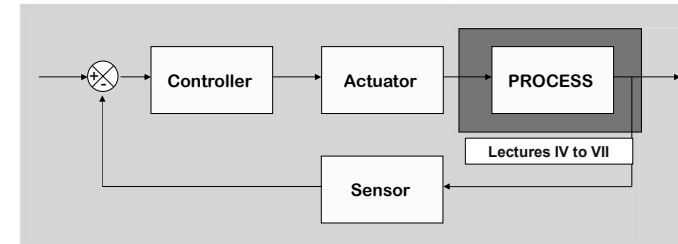
Fall 2021
Dept. of Chemical and Biological Engineering
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THE RATIONALE FOR MATHEMATICAL MODELING

- **Where to use**
 - To improve understanding of the process
 - To train plant operating personnel
 - To design the control strategy for a new process
 - To select the controller setting
 - To design the control law
 - To optimize process operating conditions
- **A Classification of Models**
 - Theoretical models (based on physicochemical law)
 - Empirical models (based on process data analysis)
 - Semi-empirical models (combined approach)

Road Map of the Lecture IV

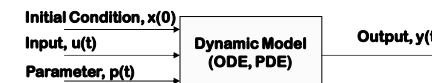
- **Basics of Process Modeling**



- Mathematical Modeling
- Steady-state model vs. Dynamic model
- Degree of freedom analysis
- Models of representative processes, etc.

DYNAMIC VERSUS STEADY-STATE MODEL

- **Dynamic model**
 - Describes time behavior of a process
 - Changes in input, disturbance, parameters, initial condition, etc.
 - Described by a set of differential equations
 - ordinary (ODE), partial (PDE), differential-algebraic(DAE)



- **Steady-state model**
 - Steady state: No further changes in all variables
 - No dependency in time: No transient behavior
 - Can be obtained by setting the time derivative term zero

MODELING PRINCIPLES

- **Conservation law**
 - Within a defined system boundary (control volume)

$$\begin{aligned} \left[\begin{array}{c} \text{rate of} \\ \text{accumulation} \end{array} \right] &= \left[\begin{array}{c} \text{rate of} \\ \text{input} \end{array} \right] - \left[\begin{array}{c} \text{rate of} \\ \text{output} \end{array} \right] \\ &+ \left[\begin{array}{c} \text{rate of} \\ \text{generation} \end{array} \right] - \left[\begin{array}{c} \text{rate of} \\ \text{disappearance} \end{array} \right] \end{aligned}$$

- **Mass balance (overall, components)**
- **Energy balance**
- **Momentum or force balance**
- **Algebraic equations: relationships between variables and parameters**

MODELING APPROACHES

- **Theoretical Model**
 - Follow conservation laws
 - Based on physicochemical laws
 - Variables and parameters have physical meaning
 - Difficult to develop
 - Can become quite complex
 - Extrapolation is valid unless the physicochemical laws are invalid
 - Used for optimization and rigorous prediction of the process behavior
- **Empirical model**
 - Based on the operation data
 - Parameters may not have physical meaning
 - Easy to develop
 - Usually quite simple
 - Requires well designed experimental data
 - The behavior is correct only around the experimental condition
 - Extrapolation is usually invalid
 - Used for control design and simplified prediction model

DEGREE OF FREEDOM (DOF) ANALYSIS

- **DOF**
 - Number of variables that can be specified independently
 - $N_F = N_V - N_E$
 - N_F : Degree of freedom (no. of independent variables)
 - N_V : Number of variables
 - N_E : Number of equations (no. of dependent variables)
 - Assume no equation can be obtained by a combination of other equations
- **Solution depending on DOF**
 - If $N_F = 0$, the system is *exactly determined*. Unique solution exists.
 - If $N_F > 0$, the system is *underdetermined*. Infinitely many solutions exist.
 - If $N_F < 0$, the system is *overdetermined*. No solutions exist.

LINEAR VERSUS NONLINEAR MODELS

- **Superposition principle**

$\forall \alpha, \beta \in \mathfrak{R}$, and for a linear operator, L

Then $L(\alpha x_1(t) + \beta x_2(t)) = \alpha L(x_1(t)) + \beta L(x_2(t))$
- **Linear dynamic model: superposition principle holds**

$\forall \alpha, \beta \in \mathfrak{R}, u_1(t) \rightarrow y_1(t)$ and $u_2(t) \rightarrow y_2(t)$

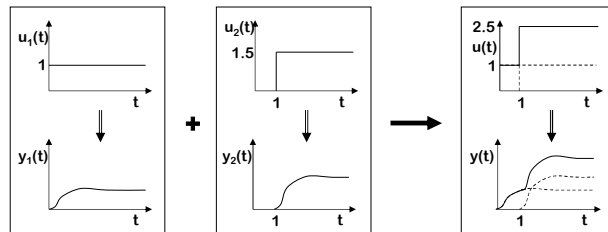
$\alpha u_1(t) + \beta u_2(t) \rightarrow \alpha y_1(t) + \beta y_2(t)$

$\forall \alpha, \beta \in \mathfrak{R}, x_1(0) \rightarrow y_1(t)$ and $x_2(0) \rightarrow y_2(t)$

$\alpha x_1(0) + \beta x_2(0) \rightarrow \alpha y_1(t) + \beta y_2(t)$

 - Easy to solve and analytical solution exists.
 - Usually, locally valid around the operating condition
- **Nonlinear: “Not linear”**
 - Usually, hard to solve and analytical solution does not exist.

ILLUSTRATION OF SUPERPOSITION PRINCIPLE



- Valid only for linear process
 - For example, if $y(t)=u(t)^2$, $(u_1(t)+1.5u_2(t))^2$ is not same as $u_1(t)^2+1.5u_2(t)^2$.

TYPICAL LINEAR DYNAMIC MODEL

Linear ODE

$$\tau \frac{dy(t)}{dt} = -y(t) + Ku(t) \quad (\tau \text{ and } K \text{ are constant, 1st order})$$

$$\frac{d^n y(t)}{dt^n} + a_{n-1} \frac{d^{n-1} y(t)}{dt^{n-1}} + \dots + a_0 y(t)$$

$$= b_m \frac{d^m u(t)}{dt^m} + b_{m-1} \frac{d^{m-1} u(t)}{dt^{m-1}} + \dots + b_0 u(t) \quad (\text{nth order})$$

Nonlinear ODE

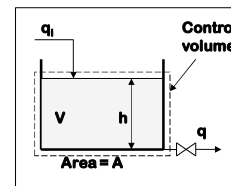
$$\tau \frac{dy(t)}{dt} = -y(t)^2 + Ku(t) \quad \tau \frac{dy(t)}{dt} = -y(t) \sin(y) + Ku(t)$$

$$\tau \frac{dy(t)}{dt} = -y(t) + K\sqrt{u(t)} \quad \tau \frac{dy(t)}{dt} = -e^{-y(t)} + Ku(t)$$

MODELS OF REPRESENTATIVE PROCESSES

Liquid storage systems

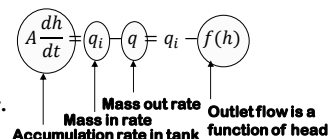
- System boundary: storage tank
- Mass in: q_i (vol. flow, indep. var)
- Mass out: q (vol. flow, dep. var)
- No generation or disappearance (no reaction or leakage)
- No energy balance



- DOF=2 (h, q_i) - 1=1

- If $f(h) = h/R_V$, the ODE is linear. (R_V is the resistance to flow)

- If $f(h) = C_V \sqrt{\rho g h / g_c}$, the ODE is nonlinear. (C_V is the valve constant)



Continuous Stirred Tank Reactor (CSTR)

- Liquid level is constant (No acc. in tank)
- Constant density, perfect mixing
- Reaction: $A \rightarrow B$ ($r = k_0 \exp(-E/RT)c_A$)
- System boundary: CSTR tank
- Component mass balance

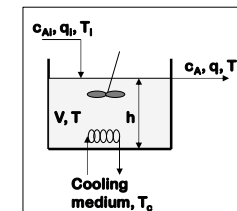
$$V \frac{dc_A}{dt} = q(c_{A_i} - c_A) - Vkc_A$$

- Energy balance

$$V\rho C_p \frac{dT}{dt} = q\rho C_p(T_i - T) + (-\Delta H)Vkc_A + UA(T_c - T)$$

- DOF analysis

- No. of variables: 6 ($q, c_A, c_{A_i}, T_p, T, T_c$)
- No. of equation: 2 (two dependent vars.: c_A, T)
- DOF=6-2=4
- Independent variables: 4 (q, c_{A_i}, T_p, T_c)
- Parameters: kinetic parameters, V, U, A , other physical properties
- Disturbances: any of q, c_{A_i}, T_p, T_c , which are not manipulatable



STANDARD FORM OF MODELS

From the previous example

$$\frac{dc_A}{dt} = \frac{q}{V}(c_{Ai} - c_A) - kc_A = f_1(c_A, T, q, c_{Ai})$$

$$\frac{dT}{dt} = \frac{q}{V}(T_i - T) + \frac{q}{\rho C_p}(-\Delta H)kc_A + \frac{UA}{\rho C_p}(T_c - T) = f_2(c_A, T, q, T_c, T_i)$$

• State-space model

$$\dot{\mathbf{x}} = d\mathbf{x}/dt = \mathbf{f}(\mathbf{x}, \mathbf{u}, \mathbf{d})$$

$$\text{where } \mathbf{x} = [x_1, \dots, x_n]^T, \mathbf{u} = [u_1, \dots, u_m]^T, \mathbf{d} = [d_1, \dots, d_l]^T$$

- **x: states**, $[c_A \ T]^T$
- **u: inputs**, $[q \ T_c]^T$
- **d: disturbances**, $[c_{Ai} \ T_i]^T$
- **y: outputs** – can be a function of above, $\mathbf{y} = \mathbf{g}(\mathbf{x}, \mathbf{d}, \mathbf{u})$, $[c_A \ T]^T$
- **If higher order derivatives exist, convert them to 1st order.**

CONVERT TO 1ST-ORDER ODE

• Higher order ODE

$$\frac{d^n x(t)}{dt^n} + a_{n-1} \frac{d^{n-1} x(t)}{dt^{n-1}} + \dots + a_0 x(t) = b_0 u(t)$$

• Define new states

$$x_1 = x, x_2 = \dot{x}, x_3 = \ddot{x}, \dots, x_n = x^{(n-1)}$$

• A set of 1st-order ODE's

$$\dot{x}_1 = x_2$$

$$\dot{x}_2 = x_3$$

⋮

$$\dot{x}_n = -a_{n-1}x_n - a_{n-2}x_{n-1} - \dots - a_0x_1 + b_0u$$

SOLUTION OF MODELS

• ODE (state-space model)

- **Linear case:** find the analytical solution via Laplace transform, or other methods.
- **Nonlinear case:** analytical solution usually does not exist.
 - Use a numerical integration, such as *RK method*, by defining initial condition, time behavior of input/disturbance
 - Linearize around the operating condition and find the analytical solution

• PDE

- **Convert to ODE by discretization of spatial variables using *finite difference approximation* and etc.**

$$\frac{\partial T_L}{\partial t} = -v \frac{\partial T_L}{\partial z} + \frac{1}{\tau_{HL}}(T_w - T_L) \longrightarrow \frac{dT_L(j)}{dt} = -\frac{v}{\Delta z} T_L(j-1) - \left(\frac{v}{\Delta z} + \frac{1}{\tau_{HL}}\right) T_L(j) + \frac{1}{\tau_{HL}} T_w \quad (j = 1, \dots, N)$$

$$\frac{\partial T_L}{\partial z} \approx \frac{T_L(j) - T_L(j-1)}{\Delta z}$$

LINEARIZATION

• Equilibrium (Steady state)

- Set the derivatives as zero: $\mathbf{0} = \mathbf{f}(\bar{\mathbf{x}}, \bar{\mathbf{u}}, \bar{\mathbf{d}})$
- Overbar denotes the steady-state value and $(\bar{\mathbf{x}}, \bar{\mathbf{u}}, \bar{\mathbf{d}})$ is the equilibrium point. (could be multiple)
- Solve them analytically or numerically using *Newton method*, etc.

• Linearization around equilibrium point

- Taylor series expansion to 1st order

$$\mathbf{f}(\mathbf{x}, \mathbf{u}) = \mathbf{f}(\bar{\mathbf{x}}, \bar{\mathbf{u}}) + \left. \frac{\partial \mathbf{f}}{\partial \mathbf{x}} \right|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} (\mathbf{x} - \bar{\mathbf{x}}) + \left. \frac{\partial \mathbf{f}}{\partial \mathbf{u}} \right|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} (\mathbf{u} - \bar{\mathbf{u}}) + \dots$$

- Ignore higher order terms

- Define deviation variables: $\mathbf{x}' = \mathbf{x} - \bar{\mathbf{x}}$, $\mathbf{u}' = \mathbf{u} - \bar{\mathbf{u}}$

$$\mathbf{x}' = \left. \frac{\partial \mathbf{f}}{\partial \mathbf{x}} \right|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} \mathbf{x}' + \left. \frac{\partial \mathbf{f}}{\partial \mathbf{u}} \right|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} \mathbf{u}' = \mathbf{A}\mathbf{x}' + \mathbf{B}\mathbf{u}'$$

Jacobian		
$\frac{\partial f_1}{\partial x_1}$...	$\frac{\partial f_1}{\partial x_n}$
\vdots	\ddots	\vdots
$\frac{\partial f_n}{\partial x_1}$...	$\frac{\partial f_n}{\partial x_n}$

$$\frac{\partial \mathbf{f}}{\partial \mathbf{x}} =$$