Carbon Dioxide Induced Crystallization of PET (Polyehtylene terephthalate)

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Thermodynamics And Green Processing Laboratory, Department of Chemical Engineering, Sogang University, Seoul, Korea *Chemical R&D Center SK Chemicals ** Department of Chemical Engineering, Seoul National University of Technology

Sogang Univ.Dept.of Chem.Eng.





In order to understand,

The temperature of crystallization behavior of PET

The heat of crystallization behavior of PET

By high-pressure carbon dioxide treatment on.





Theory

PET is made from Condensation Polymerization of TPA and EG.
PET is used to various kinds of type, as method of manufacturing.
EX) Blow molding —bottles, Melt spinning —vessel Termo forming —nursing bottle ...
For the reason, its environmentally-benign Characteristics,
Supercritical CO₂ has been used.

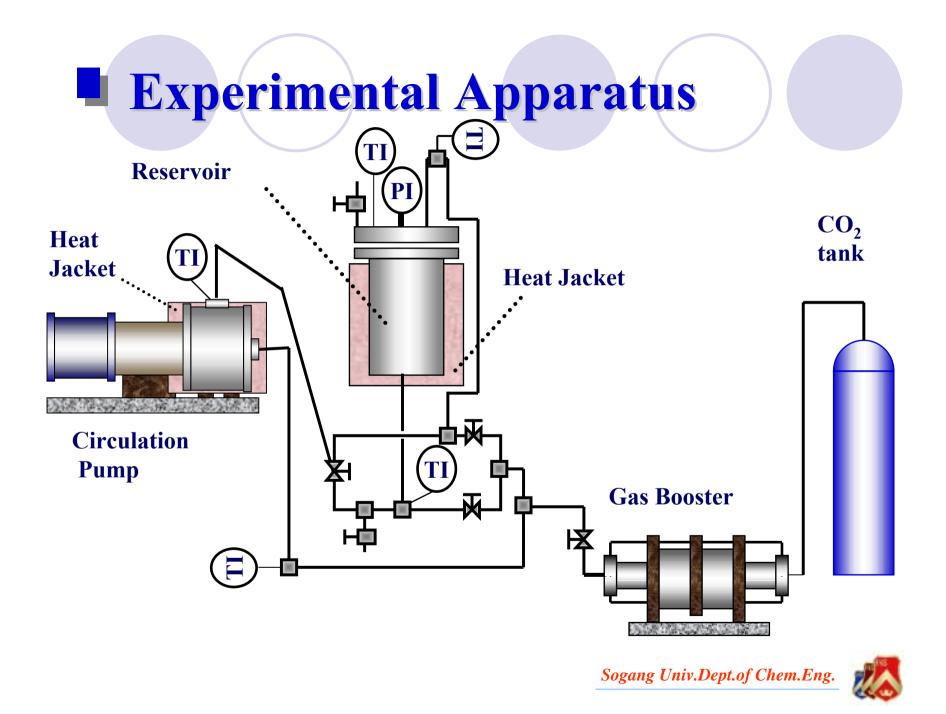
	Formula	Tg	Tc	Tm
PET	$\begin{vmatrix} 0 & 0 \\ 0 & -C & 0 \\ 0 & -C & 0 \\ 0 & -C & -CH_2 - CH_2 \\ 0 & -CH_2 - CH_2 \\ n \\ $	72	130	258



Experimental Treatment Condition with SC-CO₂

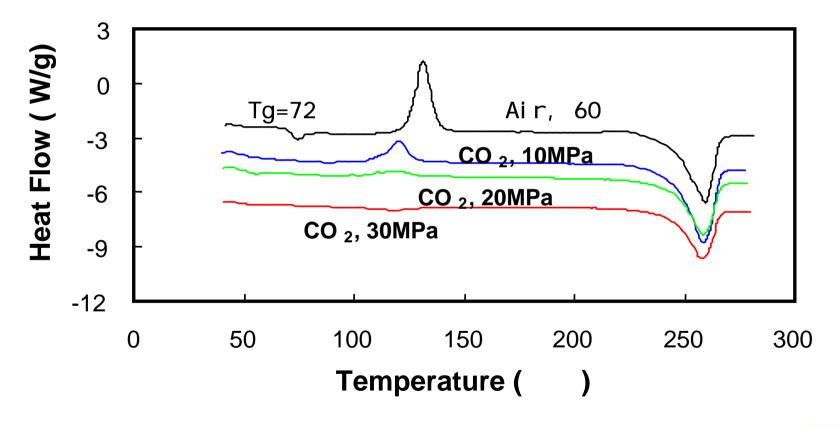
Treatment Medium	Pressure	Temperature()
Carbon Dioxide	10, 20, 30 MPa	30, 50, 60, 70
Air(Oven)	atmosphere	30, 50, 60, 70





Result

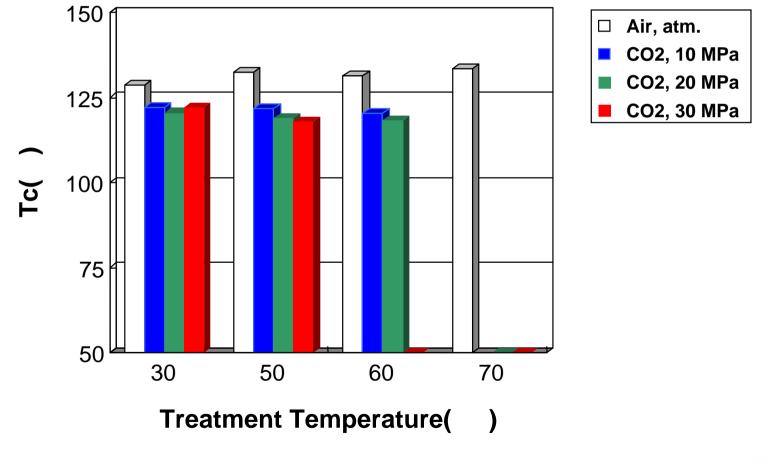
DSC Thermograms with Treatment Condition



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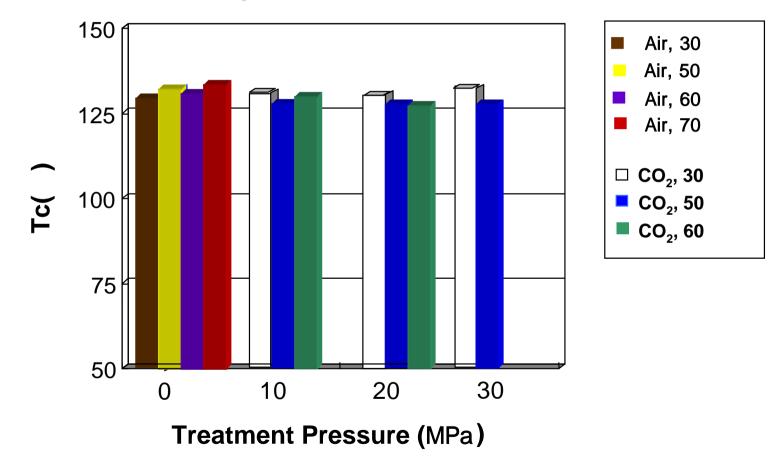


Change in Crystallization Temperature by Carbon Dioxide Treatment



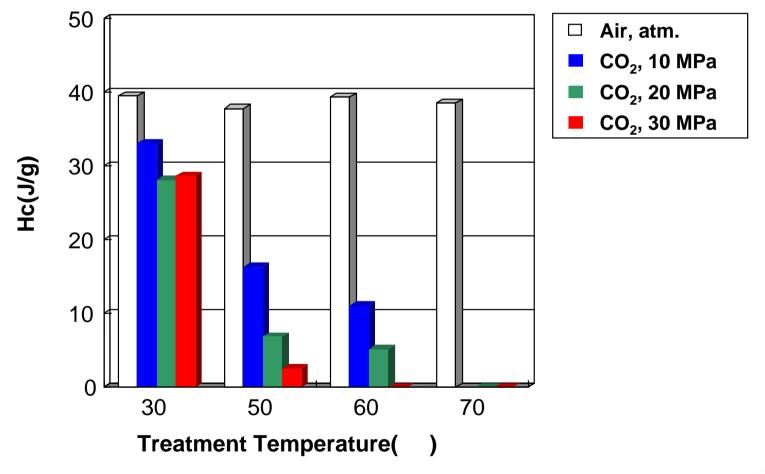


Change in Crystallization Pressure by Carbon Dioxide Treatment



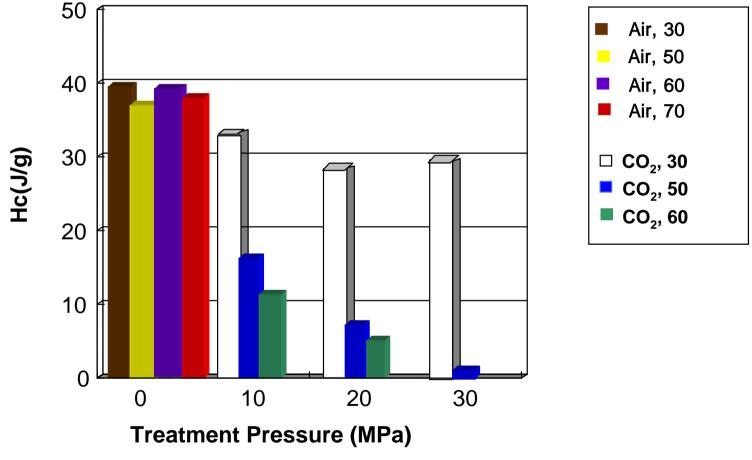


Change in Heat of Crystallization by Carbon Dioxide Treatment





Change in Heat of Crystallization by Carbon Dioxide Treatment





Concluding Remarks

CO2 treatment make <u>PET crystallize</u> <u>below the intrinsic Tc.</u>

The <u>*Hc was reduced*</u> by CO2 treatment

PET <u>crystallize more during the treatment</u>, when treated with <u>more pressure of CO2.</u>

Crystallization can be accomplished by only high pressure of CO2

Even below Tg.

