H₂S absorption with ammonia liquor from Coke Oven Gas

<u>김제영</u>*, 이성영, 김장규 포항산업과학연구원 (jykim@rist.re.kr*)

Experimental research was carried out to improve the selective solubility of H2S in Coke Oven Gas with various method.

The optimum concentrations of concentrated ammonia liquor from dissociator and crude ammonia liquor from NH3 scrubber were 18g/L and 8g/L, respectively. Changing the feeding stage of the concentrated ammonia liquor also affects absorption efficiency of H2S and it was clarified that absorption efficiency of H2S was higher when the concentrated ammonia liquor was fet to the upper stage of H2S scrubber than the lower stage.

The absorption efficiency of H2S was also affected by the flow rates of the concentrated ammonia liquor and the crude ammonia liquor. The increase of the crude ammonia liquor by 10% can improve absorption of H2S in COG.

The increase of temperature by 5° C decreases absorption of H2S by 2%. The addition of softening water into the stripped liquor that is fed to the ammonia washer #2 improves aborption of H2S, and the optimum mixing ratio of softening water to stripped liquor was 1:1.

The addition of chemical agents like NaOH and MEA into the crude ammonia liquor was also carried out.