

## The simultaneous removal of the $\text{NH}_3$ and $\text{H}_2\text{S}$ from the hot coal gases by the molybdenum-based sorbents

정석용, 김재창\*, 이수재, 박정제, 이태진<sup>1</sup>, 류창걸<sup>2</sup>  
경북대학교; <sup>1</sup>영남대학교; <sup>2</sup>한전전력연구원  
(kjchang@knu.ac.kr\*)

To investigate the simultaneous removal of  $\text{H}_2\text{S}$  and  $\text{NH}_3$ , molybdenum-based sorbents promoted with transition metals such as cobalt and nickel additive were prepared by co-precipitation method. The sulfur removing capacities and  $\text{NH}_3$  decomposition of the molybdenum-based sorbents were tested in micro reactor at 1 atm and high-temperature condition (sulfidation :  $650^\circ\text{C}$ , regeneration :  $700^\circ\text{C}$ ). The  $\text{NH}_3$  did not affect the sulfur removing capacity of molybdenum-based sorbents. The molybdenum was found to be an active component in the  $\text{NH}_3$  decomposition as well as  $\text{H}_2\text{S}$  absorption, while the support component such as  $\text{Al}_2\text{O}_3$  did not show any activity in the  $\text{NH}_3$  decomposition reaction. The removal efficiencies of the  $\text{NH}_3$  and  $\text{H}_2\text{S}$  of the molybdenum-based sorbents were 95% and 99%, respectively. The  $\text{NH}_3$  was decomposed until the breakthrough point of the  $\text{H}_2\text{S}$  removal, and the activity of the sorbent was dramatically decreased after that point.