

Synthesis and Characteristics of Morphology Controlled Titanium Dioxide for Dye-Sensitized Solar Cells

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TiO₂ nanotubes are made by treating TiO₂ powder with NaOH followed by hydrothermal treatment. The shape and structure of the TiO₂ nanotubes are determined by the precursor, number of moles of NaOH, temperature, reaction time, and washing process. In this research, TiO₂ nanotubes that crystallize at low temperatures were synthesized and their structural characteristics were examined.

After a TiO₂ sol was made via the sol-gel method, it was subjected to hydrothermal conditions to increase the specific surface area and obtain high purity TiO₂ by regulating the particle size and shape of the nanotubes.

H₂Ti₃O₇ appeared at 180 °C. The structure of the nanotubes was affected by the hydrothermal reaction temperature. The TiO₂ particles turned into TiO₂ nanotubes due to the condensation polymerization of OH⁻ and H⁺, which were weakly bound. The NaOH concentration of 10M was the most effective for the synthesis of TiO₂ nanotubes.