

Effect of Oxidizer and Alkali Metal Salts on Decomposition of Ionic Liquid of Low Concentration

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N^1, N^1, N^2, N^2 -tetramethylethane-1,2-diamine based ionic salts(TMEDA), $N^1, N^1, N^1, N^2, N^2, N^2$ -hexamethylethane-1,2-diaminium dicyanamide(HMEDA-(DCA)₂) was prepared following the well-known chemical reaction scheme. The chemical structure of the HMEDA-(DCA)₂ was confirmed using ¹³C NMR spectrum and elemental analysis. The ignition delay of 40 wt% HMEDA-(DCA)₂ solution was controlled to 20-100 msec dramatically using alkali metal salts, Li(CH₃COO), Mg(CH₃COO) and Ca(CH₃COO) as a co-catalyst when white fume nitric acid was utilized as an oxidizer and also the oxidizer of low toxicity such as hydrogen peroxide was utilized.