## Composite membranes for CO<sub>2</sub> separation

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 ${\rm CrO_3}$  particles were synthesized via a thermal reaction at 85 °C using the ionic liquid. Poly(ethylene oxide) (PEO) composite membrane was prepared for  ${\rm CO_2/N_2}$  separation with  ${\rm CrO_3}$  particles and BMIM-BF<sub>4</sub>. It was expected that the  ${\rm CrO_3}$  particles could enhance the solubility of  ${\rm CO_2}$  gas molecules. Furthermore, free imidazolium ions of BMIM-BF<sub>4</sub> played a role as agents for enhancing  ${\rm CO_2}$  transport. The permeance and selectivity of a PEO membrane without  ${\rm CrO_3}$  particles were 11.0 gas permeance units (GPU) and 6.5, respectively, while the  ${\rm CO_2}$  permeance of the composite membrane containing  ${\rm CrO_3}$  was 144 GPU with selectivity of 30. These results were attributable to the fabricated  ${\rm CrO_3}$  particles to improve the solubility of  ${\rm CO_2}$ , leading to high efficiency in  ${\rm CO_2/N_2}$  separation. Thus, the permeance and selectivity increased due to the synergistic effect of the increased  ${\rm CO_2}$  solubility by chromate esters generated from  ${\rm CrO_3}$  and the barrier effect on  ${\rm N_2}$  molecules by particles.