## Characteristics of NH<sub>3</sub> and N<sub>2</sub>O formation over Pt substituted commercial three-way catalysts (TWCs)

<u>김도영</u>, 배우빈, 변상우, 윤달영<sup>1</sup>, 정창호<sup>1</sup>, 강성봉<sup>†</sup> 광주과학기술원; <sup>1</sup>현대자동차 (sbkang@gist.ac.kr<sup>†</sup>)

An acute shortage of Pd has pushed its price to the highest level compared to Pt. From a standpoint of commercial application in the aftertreatment system, it is highly desired to identify the TWC performance by the partial substitution of Pt in Pd/Rh TWC while maintaining a constant total loading of PGM. This study characterized formations of byproducts,  $N_2O$  and  $NH_3$  along with the TWC reaction converting CO, HCs and NO. We compared six aged commercial TWCs consisting of the Pd/Rh catalysts and partially substituted Pt TWCs with respect to the location of the catalytic converter (WCC front, WCC rear and UCC). All six TWCs showed nearly 100% conversion of NO to  $N_2$  at high temperature (500 °C), regardless of the presence of Pt. However, in the lower temperature range (200 ~ 400 °C), formations of  $N_2O$  and  $NH_3$  were strongly affected by the presence of Pt. Pt-substituted Pt/Rh/Pd catalysts showed a superior NO conversion and  $N_2$  selectivity compared to the Pd/Rh TWC, regardless of the converter location. This study provides possible application of the Pt substituted TWC with comparable performances in a catalytic converter with reduced cost replacing conventional Pd/Rh based TWC.