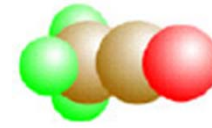


Applied Statistical Mechanics
Lecture Note - 13

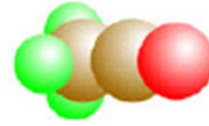


고려대학교

Molecular Dynamics Simulation

고려대학교
화공생명공학과
강정원

Contents

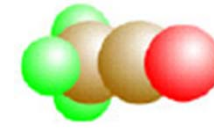


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- I. Basic Molecular Dynamics Simulation Method
- II. Properties Calculations in MD
- III. MD in Other Ensembles

I. Basic MD Simulation

- MC vs. MD



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■ MC

- Probabilistic simulation technique
- Limitations

- require the knowledge of an equilibrium distribution
- rigorous sampling of large number of possible phase-space
- gives only configurational properties (not dynamic properties !)

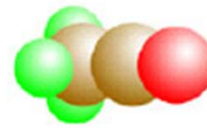
$$\langle A \rangle = \frac{\int dr^N A(r^N) \exp\{-\beta U(r^N)\}}{\int dr^N \exp\{-\beta U(r^N)\}}$$

■ MD

- Deterministic simulation technique
- Fully numerical formalism
 - numerical solution of N-body system

I. Basic MD Simulation

- The Idea

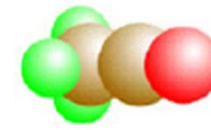


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- Follow the exactly same procedure as real experiments
 - Prepare sample
 - prepare N particles
 - solve equation of motions
 - Connect sample to measuring instruments (e.g. thermometer, viscometer,...)
 - after equilibration time, actual measurement begins
 - Measure the property of interest for a certain time interval
 - average properties
- Example : measurement of temperature

$$\left\langle \frac{1}{2} m v_{\alpha}^2 \right\rangle = \frac{1}{2} k_B T \quad \longrightarrow \quad T(t) = \sum_i \frac{m_i v_i^2(t)}{k_B N_f}$$

I. Basic MD Simulation - Equation of Motion



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- Classical Newton's equation of motion
 - Three formulation
 - Newtonian
 - Lagrangian
 - Hamiltonian
 - Hamiltonian preferred for many-body systems
 - solution of 2N differential equations

$$\frac{\partial \mathbf{r}_i}{\partial t} = \frac{\mathbf{p}_i}{m_i}$$

$$\mathbf{r} = \mathbf{r}(r_x, r_y, r_z)$$

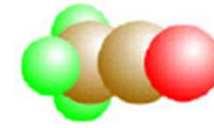
$$\mathbf{p} = \mathbf{p}(p_x, p_y, p_z) \longrightarrow$$

$$\frac{\partial \mathbf{p}_i}{\partial t} = \mathbf{F}_i$$

$$\mathbf{F}_i = \sum_{\substack{j=1 \\ j \neq i}} \mathbf{F}_{ij}$$

Solution methods : Finite Difference Method

I. Basic MD Simulation - Verlet Algorithm



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- Verlet (1967) : Very simple, efficient and popular algorithm

$$\mathbf{r}(t + \delta t) = \mathbf{r}(t) + \frac{1}{m}\mathbf{p}(t)\delta t + \frac{1}{2m}\mathbf{F}(t)\delta t^2 + \frac{1}{3!}\ddot{\mathbf{r}}(t)\delta t^3 + O(\delta t^4)$$

$$\mathbf{r}(t - \delta t) = \mathbf{r}(t) - \frac{1}{m}\mathbf{p}(t)\delta t + \frac{1}{2m}\mathbf{F}(t)\delta t^2 - \frac{1}{3!}\ddot{\mathbf{r}}(t)\delta t^3 + O(\delta t^4)$$



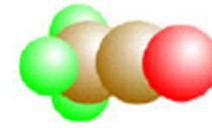
$$\mathbf{r}(t + \delta t) + \mathbf{r}(t - \delta t) = 2\mathbf{r}(t) + \frac{1}{m}\mathbf{F}(t)\delta t^2 + O(\delta t^4)$$



$$\mathbf{r}(t + \delta t) = 2\mathbf{r}(t) - \mathbf{r}(t - \delta t) + \frac{1}{m}\mathbf{F}(t)\delta t^2 + O(\delta t^4)$$

feature : update without calculating momentum (p)

I. Basic MD Simulation - Leapfrog Algorithm



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- Hockeny (1970), Potter (1972)
- Half-step leap-frog algorithm
- Mathematically equivalent to Verlet algorithm

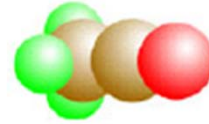
$$\mathbf{r}(t + \delta t) = \mathbf{r}(t) + \mathbf{v}(t + \frac{1}{2} \delta t) \delta t$$

$$\mathbf{v}(t + \frac{1}{2} \delta t) = \mathbf{v}(t - \frac{1}{2} \delta t) + \frac{1}{m} \mathbf{F}(t) \delta t$$



$$\mathbf{r}(t + \delta t) = \mathbf{r}(t) + \left[\mathbf{v}(t - \frac{1}{2} \delta t) + \frac{1}{m} \mathbf{F}(t) \delta t \right] \delta t$$

2. Properties Calculation in MD - Energies

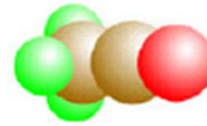


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- Potential energy
 - Can be calculated during force calculation
- Kinetic energy

$$K = \sum \frac{1}{2} m v_i^2$$

2. Properties Calculation in MD - Pressures

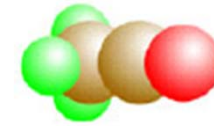


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- In an MD simulation, calculation of pressure using tensor notation is not the most efficient method.
- For homogeneous systems, there is simple way to calculate pressure (Irving and Kirkwood, 1950)

$$\begin{aligned}
 & \text{Kinetic – ideal gas term} && \text{Configurational – called “Virial”} \\
 & \downarrow && \downarrow \\
 P = \frac{1}{V} & \left[\sum_i m_i \mathbf{v}_i(t) \mathbf{v}_i(t) + \sum_i \sum_j \mathbf{r}_{ij}(t) \mathbf{F}_{ij}(t) \right] \\
 = \rho k_B T & + \frac{1}{2V} \sum_i \sum_j \mathbf{r}_{ij}(t) \mathbf{F}_{ij}(t) && \begin{array}{l} \nearrow \text{Calculate when force update} \\ \downarrow \text{Calculate when velocity update} \end{array}
 \end{aligned}$$

2. Properties Calculation in MD - Transport Properties



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■ Approaches for transport properties

□ Method 1 : *NEMD* (Non-equilibrium Molecular Dynamics)

- Continuous addition and removal of conserved quantities
- Gives high signal-to-noise ratio (good statistics)

□ Method 2 : Equilibrium molecular dynamics

- Start with anisotropic configuration of mass, momentum and energy
- Observe natural fluctuations and dissipation of mass, momentum and energy
- Poor signal-to-noise ration (poor statistics)
- All transport properties can be measured at once

2. Properties Calculation in MD - Transport Properties



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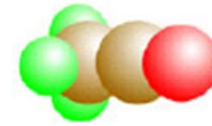
■ Differential Balance Equation

Mass	Energy	Momentum
$\frac{\partial c(r,t)}{\partial t} + \nabla \cdot \mathbf{j} = 0$	$c_p \frac{\partial T(r,t)}{\partial t} + \nabla \cdot \mathbf{q} = 0$	$\rho \frac{D\mathbf{v}(r,t)}{Dt} + \nabla \cdot \underline{\underline{\boldsymbol{\tau}}} = 0$

■ Constitutive Equations

Fick's Law	Fourier's Law	Newton's Law
$\mathbf{j} = -D\nabla c$	$q = -k\nabla T$	$\tau_{xy} = -\nu \nabla_y (\rho v_x)$

2. Properties Calculation in MD - Transport Properties



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- Purpose : Obtain transport coefficient by molecular simulation
 - Not that the “laws” are only approximation that apply not-too-large gradients
 - In principle transfer coefficients depends on c , T and v

- **Green-Kubo Relation**
 - Relation between transport properties and integral over time-correlation function.

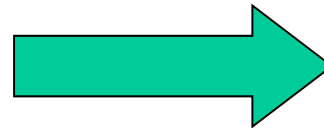
2. Properties Calculation in MD - Transport Properties



- Consider self-diffusion in a pure substance
- Consider how molecules are dissipated when initial configurations are given as Dirac delta function
- Combine mass balance eqn. With Fick's Law

$$\frac{\partial c(\mathbf{r}, t)}{\partial t} - D \nabla^2 c(\mathbf{r}, t) = 0$$

$$\text{B.C. } c(\mathbf{r}, t) = \delta(\mathbf{r})$$



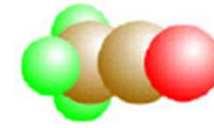
Solution

$$c(\mathbf{r}, t) = (2\pi Dt)^{-2/d} \exp\left(-\frac{r^2}{2Dt}\right)$$

Dimensionality
of given system



2. Properties Calculation in MD - Transport Properties



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We do not need concentration itself $c(\mathbf{r},t)$ - just diffusion coefficient (D)

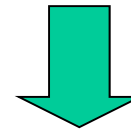
$$\langle r^2(t) \rangle = \int c(\mathbf{r},t) r^2 dr$$

$$\int c(\mathbf{r},t) dr = 1$$

$$\frac{\partial c(\mathbf{r},t)}{\partial t} - D \nabla^2 c(\mathbf{r},t) = 0$$

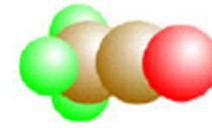


$$\frac{\partial \langle r^2(t) \rangle}{\partial t} = 2dD$$



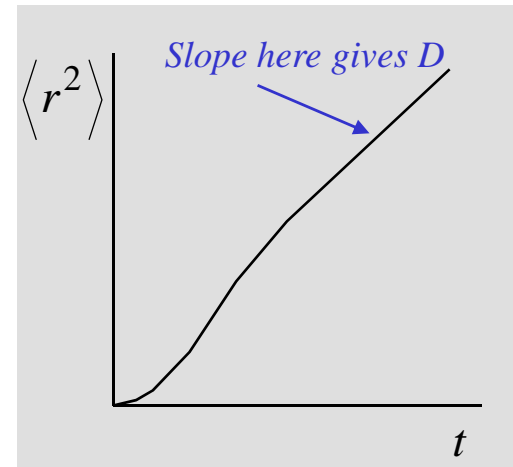
$$\langle r^2(t) \rangle = \frac{1}{N} \sum (r_i(t))^2 = 2dDt$$

2. Properties Calculation in MD - Transport Properties



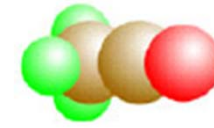
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$$\langle r^2(t) \rangle = \frac{1}{N} \sum (r_i(t))^2 = 2dDt$$



- *Plot of t vs. square of traveled distance gives diffusion coefficient*
- *In 3D -space, $\langle r^2 \rangle$ is mean square displacement (MSD)*

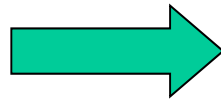
2. Properties Calculation in MD - Transport Properties



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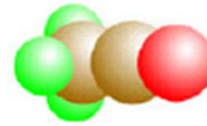
- An alternative formulation using velocity instead of particle position

$$\mathbf{r}_i(t) = \int_0^t \mathbf{v}(\tau) d\tau$$



$$\begin{aligned}\langle r^2(t) \rangle &= \left\langle \int_0^t \mathbf{v}(\tau_1) d\tau_1 \cdot \int_0^t \mathbf{v}(\tau_2) d\tau_2 \right\rangle \\ &= \int_0^t d\tau_1 \int_0^t d\tau_2 \langle \mathbf{v}(\tau_2) \cdot \mathbf{v}(\tau_1) \rangle \\ &= 2 \int_0^t d\tau_1 \int_0^{\tau_1} d\tau_2 \langle \mathbf{v}(\tau_2) \cdot \mathbf{v}(\tau_1) \rangle \\ &= 2 \int_0^t d\tau_1 \int_0^{\tau_1} d\tau_2 \langle \mathbf{v}(0) \cdot \mathbf{v}(\tau_1 - \tau_2) \rangle \\ &= 2 \int_0^t d\tau_1 \int_0^{\tau_1} d\tau \langle \mathbf{v}(0) \cdot \mathbf{v}(\tau) \rangle \\ 2dDt &= 2t \int_0^t d\tau \langle \mathbf{v}(0) \cdot \mathbf{v}(\tau) \rangle\end{aligned}$$

2. Properties Calculation in MD - Transport Properties

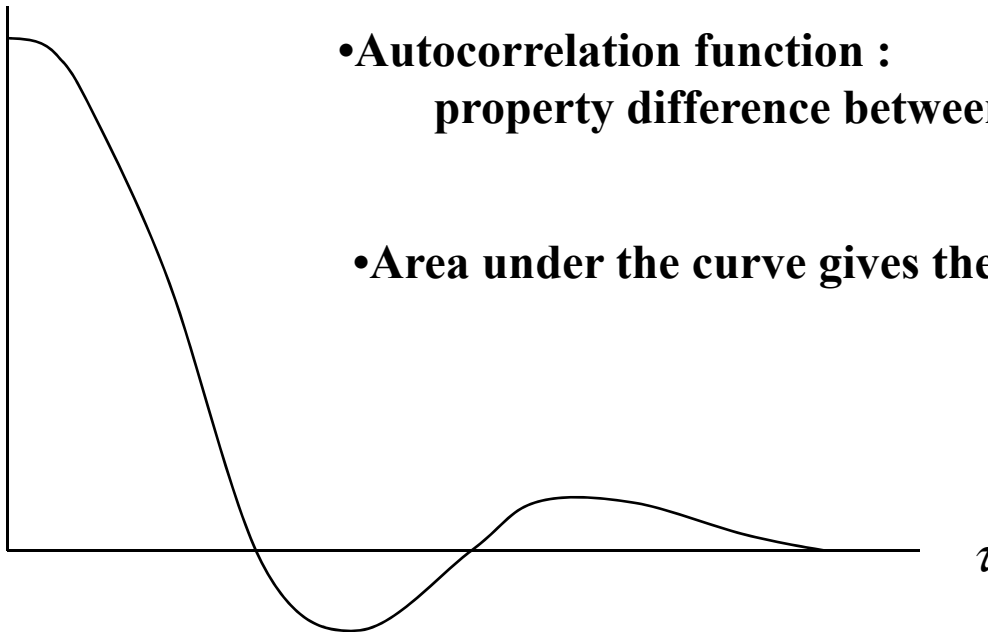


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$$2dDt = 2t \int_0^t d\tau \langle \mathbf{v}(0)\mathbf{v}(\tau) \rangle \xrightarrow{t \rightarrow \infty} D = \frac{1}{d} \int_0^{\infty} d\tau \langle \mathbf{v}(0)\mathbf{v}(\tau) \rangle$$

$\langle \mathbf{v}(0)\mathbf{v}(\tau) \rangle$

$\langle \mathbf{v}(0)\mathbf{v}(\tau) \rangle = \langle \mathbf{v}(t')\mathbf{v}(t'') \rangle$

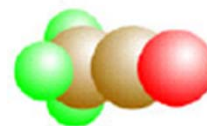


• **Autocorrelation function :**
property difference between two adjacent time steps

• **Area under the curve gives the value of self-diffusion coefficient**

2. Properties Calculation in MD

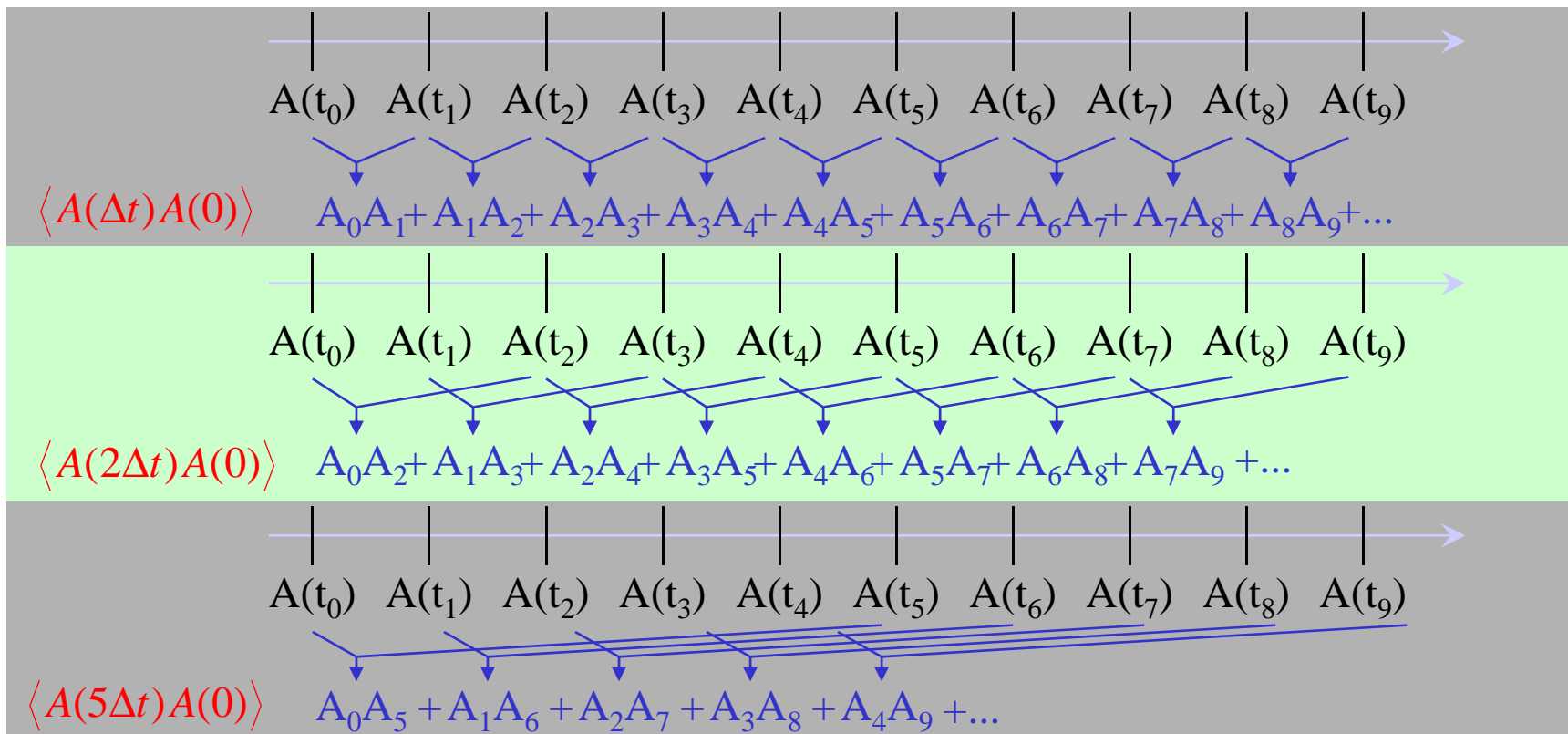
- Evaluation of time correlation functions



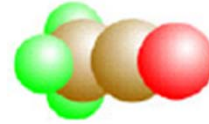
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■ Time consuming and require a lot of storage

- Alternative method : FFT (Fast Fourier Transform), Coarse Graining method



2. Properties Calculation in MD - Transport Properties



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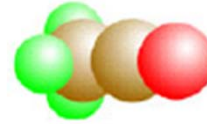
■ Zero-shear viscosity

$$\eta = \frac{1}{VkT} \int_0^{\infty} d\tau \langle \sigma_{xy}(0) \sigma_{xy}(\tau) \rangle \quad \sigma_{xy} = \sum_i \left[m_i v_i^x v_i^y + \frac{1}{2} \sum_{i \neq j} x_{ij} f_y(r_{ij}) \right]$$

■ Thermal Conductivity

$$\lambda_T = \frac{1}{VkT^2} \int_0^{\infty} d\tau \langle q(0)q(\tau) \rangle \quad q = \frac{d}{dT} \sum_i \left[m_i v_i^2 + \frac{1}{2} \sum_{i \neq j} u(r_{ij}) \right]$$

2. Properties Calculation in MD - Radial Distribution Function

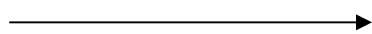
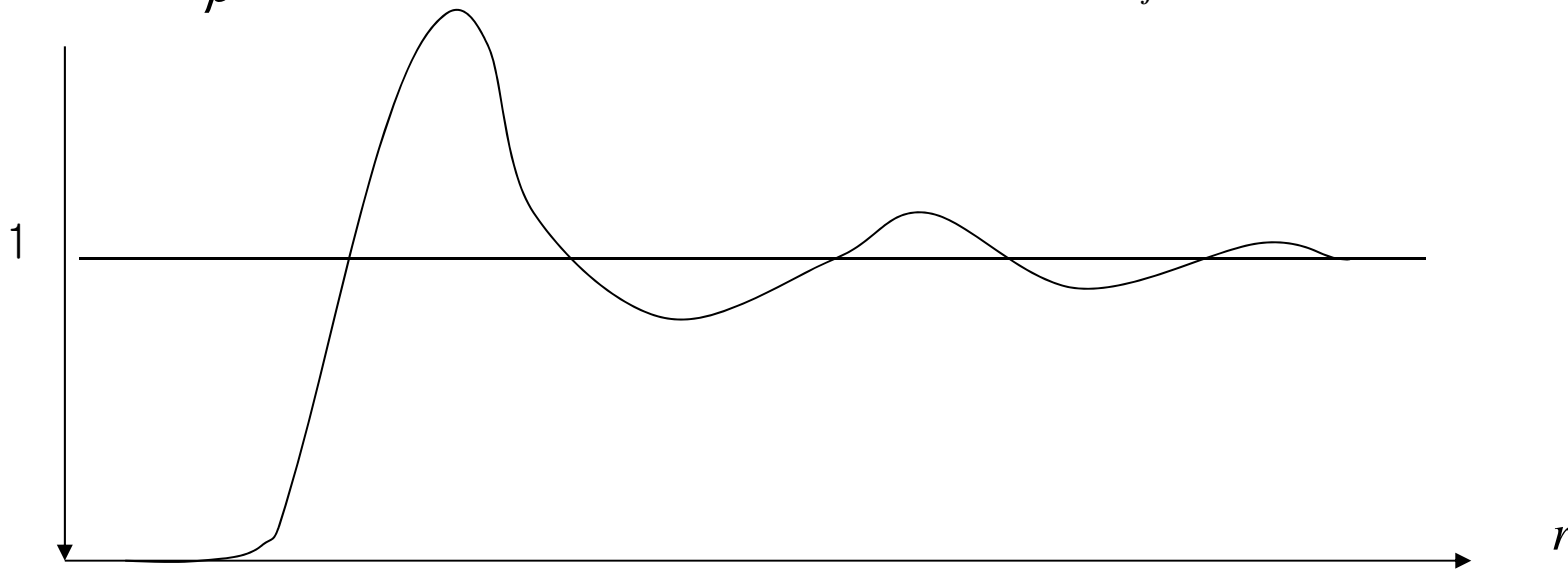


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- Time averaged value of number density
- Ensemble averaged number density

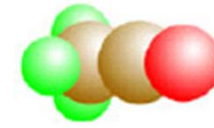
$$g(r) = \frac{\rho(r)}{\rho}$$

$$g(r) = \frac{V}{N} \left\langle \sum_i \sum_{j \neq i} \delta(\mathbf{r} - \mathbf{r}_{ij}) \right\rangle$$



Just count the number of molecules within a range

3. MD in Other Ensembles - Constraints



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- With proper choice of $g(r)$, we can calculate useful thermodynamic properties

- Internal energy $U^c = 2\pi N\rho \int_0^\infty \phi(r) g(r) r^2 dr$

- Pressure $P = \rho kT - \frac{2\pi\rho^2}{3} \int_0^\infty \frac{d\phi(r)}{dr} g(r) r^3 dr$

- Chemical Potential

$$\mu = kT \ln\left(\frac{\rho\Lambda^3}{q_{\text{int}}}\right) - \frac{2\pi}{3} \left\{ \frac{\partial}{\partial N} \left[N^2 \int_V^\infty \frac{dV'}{V'^2} \int_0^\infty \frac{d\phi(r)}{dr} g(r) r^3 dr \right] \right\}$$

3. MD in Other Ensembles – Constraints



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■ Hamiltonian formulation

- Conservation of kinetic + potential energy

$$H = K + U$$

- (N, V, E) ensemble

- Cannot be applied to other ensemble

- constant T , constant P , ...
- for example we can keep const T while H is constant
- distribution of K and U

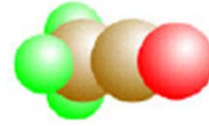
■ Two types of constraints

- Holonomic constraints : may be integrated out of equation of motion

- Nonholonomic constraints : non-integrable (involves velocities)

- Temperature, pressure, stress, ...

3. MD in Other Ensembles – Constraints



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- Force momentum temperature to remain constant
- One (bad) approach
 - at each time step scale momenta to force K to desired value
 - advance positions and momenta
 - apply $p^{\text{new}} = \lambda p$ with λ chosen to satisfy
 - repeat
 - “equations of motion” are irreversible
 - “transition probabilities” cannot satisfy detailed balance
 - does not sample any well-defined ensemble

3. MD in Other Ensembles

– Constraints



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■ “Gauss’ principle of least constraints”

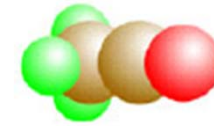
- Gaussian constraints : perturbative force introduced into the equation of motion minimizes the deviation to classical trajectories of particles from their unperturbed trajectories
- Consider a function f , a function of particle acceleration

$$f(\ddot{\mathbf{r}}_i) = \frac{1}{2} \sum_i m \left[\ddot{\mathbf{r}}_i - \frac{\mathbf{F}_i}{m_i} \right]^2$$

- $f=0$: normal Newtonian equation of motion
- otherwise, constrained non-Newtonian equation of motion
- Gauss’ principle : physical acceleration $\rightarrow f$ to be minimum

$$\frac{\partial}{\partial \ddot{\mathbf{r}}_i} (f(\ddot{\mathbf{r}}_i) - \zeta g(\ddot{\mathbf{r}}_i)) = 0 \quad \zeta : \text{Lagrangin (Gauss) Multiplier}$$

3. MD in Other Ensembles – Constraints



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■ Constant Temperature constraints

$$G(\dot{\mathbf{r}}_i, t) = \sum_i \frac{m_i \dot{\mathbf{r}}_i^2}{2} - \frac{3NkT}{2} = 0$$

$$g(\ddot{\mathbf{r}}_i, \dot{\mathbf{r}}_i, t) = \frac{dG}{dt} = \sum_i m \dot{\mathbf{r}}_i \bullet \ddot{\mathbf{r}}_i = 0$$

$$\frac{\partial}{\partial \dot{\mathbf{r}}_i} (f(\dot{\mathbf{r}}_i) - \zeta g(\dot{\mathbf{r}}_i)) = 0$$

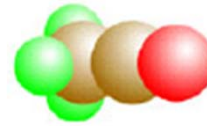
$$\frac{\partial}{\partial \dot{\mathbf{r}}_i} \left[\frac{1}{2} \sum_j m \left(\dot{\mathbf{r}}_i - \frac{\mathbf{F}_j}{m_j} \right)^2 - \sum_j m_j \dot{\mathbf{r}}_j \bullet \dot{\mathbf{r}}_j \right] = 0$$

$$m \underbrace{\ddot{\mathbf{r}}_i}_{\text{Newtonian}} = \mathbf{F}_i - \underbrace{\zeta m_i \dot{\mathbf{r}}_i}_{\text{Constraint force}}$$

Newtonian

Constraint force

3. MD in Other Ensembles – Constraints



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■ Modified equation of motion

$$\mathbf{r}_i = \frac{\mathbf{p}_i}{m_i}$$

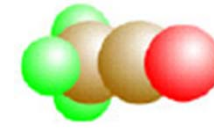
$$\mathbf{p}_i = \mathbf{F}_i - \zeta m_i \dot{\mathbf{r}}_i$$

$$\zeta = \frac{\sum_i \dot{\mathbf{r}}_i \cdot \mathbf{F}_i}{\sum_i m_i \dot{\mathbf{r}}_i^2}$$



one of good approach, but temperature is not specified !

3. MD in Other Ensembles – Nose Thermostat



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■ Extended Lagrangian Equation of Motion

$$L_{Nose} = \sum_{i=1}^N \frac{m_i (\mathbf{s}\dot{\mathbf{r}}_i)^2}{2} - U(\mathbf{r}^N) + \frac{Q}{2} \dot{s}^2 - gkT \ln s$$

$$\mathbf{p}_i \equiv \frac{\partial L}{\partial \dot{\mathbf{r}}_i} = m_i s^2 \dot{\mathbf{r}}_i$$

$$p_s \equiv \frac{\partial L}{\partial \dot{s}} = Q\dot{s}$$

$$U_s = -gkT \ln s$$

$$K_s = \frac{1}{2} Q\dot{s}^2$$

3. MD in Other Ensembles – Nose-Hoover Thermostat



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■ Equations of motion

$$\dot{\mathbf{r}}_i = \frac{\mathbf{p}_i}{m_i}$$

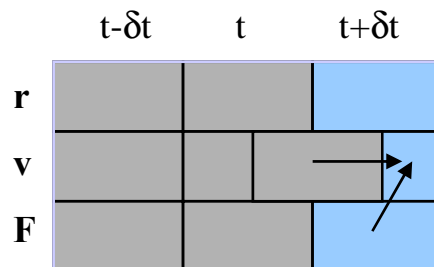
$$\dot{\mathbf{p}}_i = \mathbf{F}_i - \xi \mathbf{p}_i$$

$$\frac{\dot{s}}{s} = \xi$$

$$\dot{\xi} = \frac{1}{Q} \left(\sum_{i=1}^N \frac{p_i^2}{m_i} - gkT \right)$$

■ Integration schemes

- predictor-corrector algorithm is straightforward
- Verlet algorithm is feasible, but tricky to implement



At this step, update of ξ depends on p ; update of p depends on ξ

$$\dot{\mathbf{p}}_i = \mathbf{F}_i - \xi \mathbf{p}_i$$

$$\dot{\xi} = \frac{1}{Q} \left(\sum_{i=1}^N \frac{p_i^2}{m_i} - gkT \right)$$