

Chapter 11. Principle of heat flow in fluid

convection : Natural convection & Forced convection

1. Newton's equation of cooling

$$q \propto A \Delta T$$

$$q = h A \Delta T \quad \Delta T = \text{⊕} - \text{⊗}$$

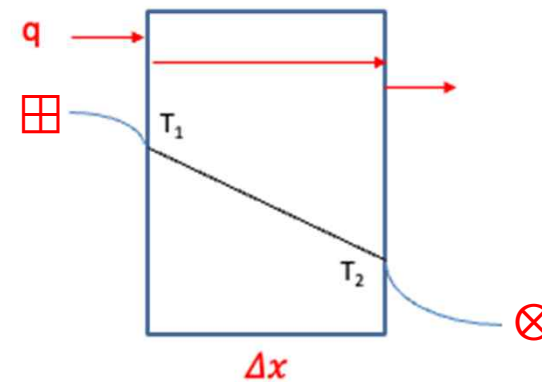
where, h is heat transfer coefficient

$h = h(\text{geometry, physical property of fluid, fluid velocity...})$

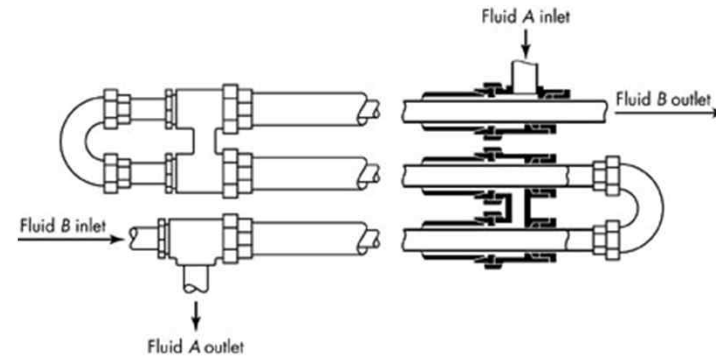
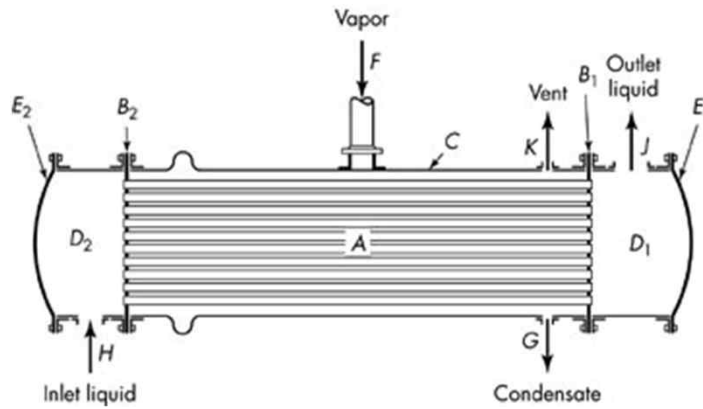
$$\left[\frac{\text{W}}{\text{m}^2 \text{°C}} \right], \left[\frac{\text{Btu}}{\text{ft}^2 \text{ hr } \text{°F}} \right]$$

$$q = \frac{\Delta T}{R_{\text{conv}}}$$

$$R_{\text{conv}} = \frac{1}{h A}$$



2. Types of Heat Exchanger (HEX)



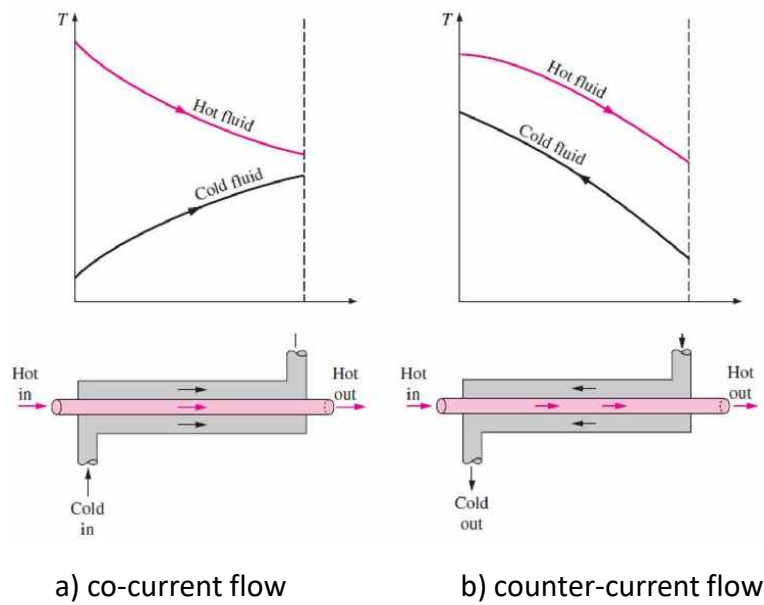
* Contact methods

Counter current flow

Co-current flow

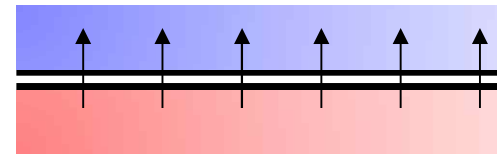
Cross current flow

Double pipe HEX



$$T_c \quad T_c + dT_c$$

$$H_c \quad H_c + dH_c$$



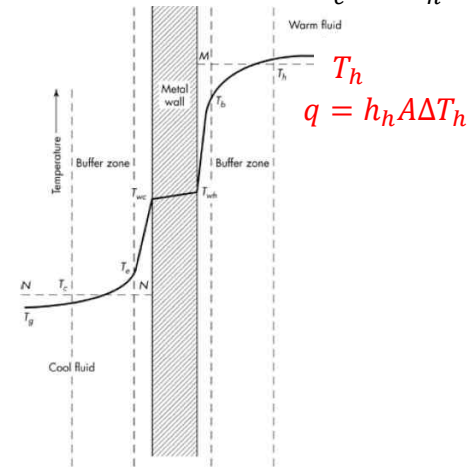
$$l \quad l + dl$$

$$T_h \quad T_h + dT_h$$

$$H_h \quad H_h + dH_h$$

$$T_c$$

$$q = h_c A \Delta T_c$$



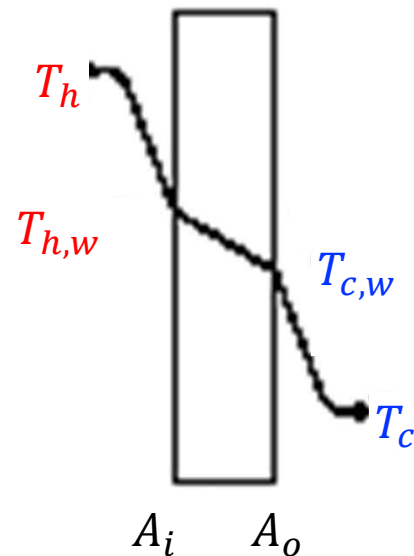
$$q = h_h A \frac{\Delta T_w}{x_w}$$

3. Rate of heat transfer

1) Overall heat transfer coefficient

<Assumptions>

- 1) \dot{m}_h, \dot{m}_c : constant
- 2) C_{ph}, C_{pc} : constant



$$q = U A (T_h - T_c)$$

$T_h - T_c$: Overall local temperature difference

U : local overall heat transfer coefficient

Differential rate of heat transfer through infinitesimal cross-sectional area, dA

$$\begin{aligned} \delta q &= U_i dA_i \Delta T \\ &= U_o dA_o \Delta T \end{aligned}$$

$$\begin{aligned} dA_i &= \pi D_i dl \\ dA_o &= \pi D_o dl \end{aligned}$$

$$U_i dA_i = U_o dA_o$$

$$\frac{U_i}{U_o} = \frac{dA_o}{dA_i} = \frac{D_o}{D_i}$$

3. Rate of heat transfer

2) Heat balance

Hot fluid side

$$H_h = (H_h + dH_h) + \delta q$$

$$dH_h = -\delta q$$

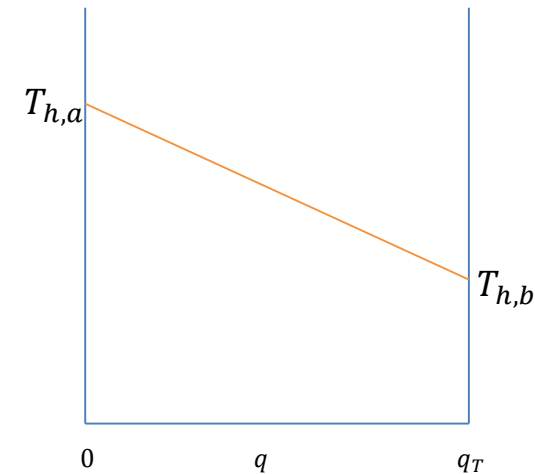
$$dH_h = \dot{m}_h C_{ph} dT_h$$

$$\delta q = -\dot{m}_h C_{ph} dT_h$$

$$q = -\dot{m}_h C_{ph} (T_h - T_{h,a}) = \dot{m}_h C_{ph} (T_{h,a} - T_h)$$

Total heat transfer rate

$$q_T = \dot{m}_h C_{ph} (T_{h,a} - T_{h,b})$$



3. Rate of heat transfer

2) Heat balance

Cold fluid side

$$(H_c + dH_c) + \delta q = H_c$$

$$H_c = -\delta q$$

$$dH_c = \dot{m}_c C_{pc} dT_c = -\delta q$$

$$\int \delta q = \int_{T_{c,a}}^{T_c} -\dot{m}_c C_{pc} dT_c$$

$$q = -\dot{m}_c C_{pc} (T_c - T_{c,a}) = \dot{m}_c C_{pc} (T_{c,a} - T_c)$$

Total heat transfer rate

$$q_T = \dot{m}_c C_{pc} (T_{c,a} - T_{c,b}) = -\Delta H_{T,h} = \Delta H_{T,c}$$

3. Rate of heat transfer

2) Heat balance

Overall heat balance

$$(q_T =) \dot{m}_c C_{pc} (T_{c,a} - T_{c,b}) = \dot{m}_h C_{ph} (T_{h,a} - T_{h,b})$$

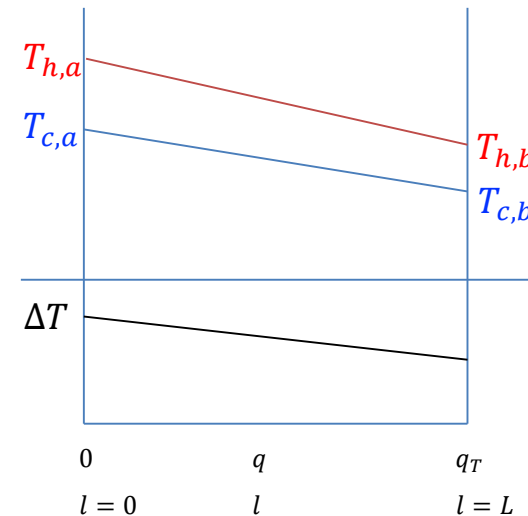
$$-\Delta H_{T,h} = \Delta H_{T,c}$$

$$-\{\dot{m}_h C_{ph} (T_{h,b} - T_{h,a})\} = \dot{m}_c C_{pc} (T_{c,a} - T_{c,b})$$

$$\text{slope} = \frac{d(\Delta T)}{\delta q} = \frac{\Delta T_2 - \Delta T_1}{q_T} \quad \delta q = U dA \Delta T$$

$$\frac{d(\Delta T)}{U dA \Delta T} = \frac{\Delta T_2 - \Delta T_1}{q_T}$$

where, $q_T = \dot{m}_c C_{pc} (T_{c,a} - T_{c,b}) = \dot{m}_h C_{ph} (T_{h,a} - T_{h,b})$



3. Rate of heat transfer

Case I, U is constant

$$\int_{\Delta T_1}^{\Delta T_2} \frac{1}{U} \frac{d(\Delta T)}{\Delta T} = \int_0^{A_T} \frac{\Delta T_2 - \Delta T_1}{q_T} dA$$

$$\frac{1}{U} \ln \frac{\Delta T_2}{\Delta T_1} = \frac{\Delta T_2 - \Delta T_1}{q_T} A_T$$

$$\therefore q_T = UA_T \frac{\Delta T_2 - \Delta T_1}{\ln \frac{\Delta T_2}{\Delta T_1}}$$

$$q_T = UA_T \overline{\Delta T_L}$$

$$\overline{\Delta T_L} = \frac{\Delta T_2 - \Delta T_1}{\ln \frac{\Delta T_2}{\Delta T_1}} \quad \text{Log mean temperature difference (LMTD)}$$

If \dot{m} and C_p are constant

$$\frac{d(\Delta T)}{\delta q} = \frac{\Delta T_2 - \Delta T_1}{q_T - 0}$$

$$\leftarrow \delta q = u d \bar{A} \Delta T$$

3. Rate of heat transfer

Case II, U is **NOT** constant

$$U = U(\Delta T) = a + b\Delta T$$

$$\frac{d(\Delta T)}{U dA \Delta T} = \frac{\Delta T_2 - \Delta T_1}{q_T}$$

$$\int_{\Delta T_1}^{\Delta T_2} \frac{a + b\Delta T - b\Delta T}{a(a + b\Delta T)\Delta T} d(\Delta T) = \int_0^{A_T} \frac{\Delta T_2 - \Delta T_1}{q_T} dA$$

$$\frac{1}{a} \int_{\Delta T_1}^{\Delta T_2} \left[\frac{a + b\Delta T}{(a + b\Delta T)\Delta T} - \frac{b\Delta T}{(a + b\Delta T)\Delta T} \right] d(\Delta T)$$

$$= \frac{1}{a} \int_{\Delta T_1}^{\Delta T_2} \left[\frac{1}{\Delta T} - \frac{b\Delta T}{(a + b\Delta T)\Delta T} \right] d(\Delta T) = \frac{\Delta T_2 - \Delta T_1}{q_T} A_T$$

$$\frac{1}{a} \left\{ \ln \frac{\Delta T_2}{\Delta T_1} - \ln \frac{a + b\Delta T_2}{a + b\Delta T_1} \right\} = \frac{\Delta T_2 - \Delta T_1}{q_T} A_T$$

$$\ln \frac{\Delta T_2(a + b\Delta T_1)}{\Delta T_1(a + b\Delta T_2)} = \frac{a\Delta T_2 - a\Delta T_1}{q_T} A_T$$

$$\ln \frac{\Delta T_2 U_1}{\Delta T_1 U_2} = \frac{\Delta T_2 U_1 - \Delta T_1 U_2}{q_T} A_T$$

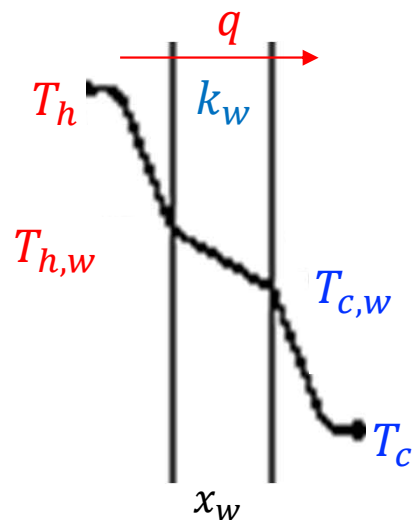
$$\begin{aligned} \Delta T_2 U_1 &= \Delta T_2 a + b\Delta T_1 \Delta T_2 \\ \Delta T_1 U_2 &= \Delta T_1 a + b\Delta T_2 \Delta T_1 \end{aligned}$$

$$\Delta T_2 U_1 - \Delta T_1 U_2 = a\Delta T_2 a - a\Delta T_1$$

$$\therefore q_T = A_T \frac{U_1 \Delta T_2 - U_2 \Delta T_1}{\ln \frac{U_1 \Delta T_2}{U_2 \Delta T_1}}$$

4. Relation between overall heat transfer coefficient and individual heat transfer coefficient

1) For clean surface (no deposit)



$$\delta q = \frac{\Delta T_i}{R_i} \rightarrow \left(\begin{array}{l} \Delta T_i = T_h - T_{h,w} \\ R_i = 1/(h_i dA_i) \end{array} \right)$$

$$\delta q = \frac{\Delta T_w}{R_w} \rightarrow \left(\begin{array}{l} \Delta T_w = T_{h,w} - T_{c,w} \\ R_i = x_w/k_w d\bar{A}_L \end{array} \right)$$

$$\delta q = \frac{\Delta T_o}{R_o} \rightarrow \left(\begin{array}{l} \Delta T_o = T_{c,w} - T_c \\ R_o = 1/(h_o dA_o) \end{array} \right)$$

$$\delta q = \frac{T_h - T_c}{R_i + R_w + R_o} = \frac{\Delta T}{R_i + R_w + R_o} \text{ ----- A}$$

$$\delta q = U_i dA_i \Delta T = U_o dA_o \Delta T \text{ ----- B}$$

4. Relation between overall heat transfer coefficient and individual heat transfer coefficient

1) For clean surface (no deposit)

From A & B,

$$\frac{1}{U_i dA_i} = R_i + R_w + R_o = \frac{1}{h_i dA_i} + \frac{x_w}{k_w d\bar{A}_L} + \frac{1}{h_o dA_o}$$

$$\frac{1}{U_i} = \frac{1}{h_i} + \frac{x_w dA_i}{k_w d\bar{A}_L} + \frac{1}{h_o} \frac{dA_i}{dA_o}$$

$$\frac{1}{U_i} = \frac{1}{h_i} + \frac{x_w D_i}{k_w \bar{D}_L} + \frac{1}{h_o} \frac{D_i}{D_o}$$

$$\begin{aligned} \delta q &= U_i dA_i \Delta T \rightarrow \frac{\Delta T}{1/(U_i dA_i)} \text{ ---- A} \\ &= U_o dA_o \Delta T \rightarrow \frac{\Delta T}{1/(U_o dA_o)} \text{ ---- C} \end{aligned}$$

$$\delta q = \frac{\Delta T}{\frac{1}{h_i dA_i} + \frac{x_w}{k_w d\bar{A}_L} + \frac{1}{h_o dA_o}}$$

$$\frac{1}{U_i} = \frac{1}{h_i} + \frac{x_w D_i}{k_w \bar{D}_L} + \frac{1}{h_o} \frac{D_i}{D_o}$$

$$\frac{1}{U_o} = \frac{1}{h_o} \frac{D_o}{D_i} + \frac{x_w D_o}{k_w \bar{D}_L} + \frac{1}{h_i}$$

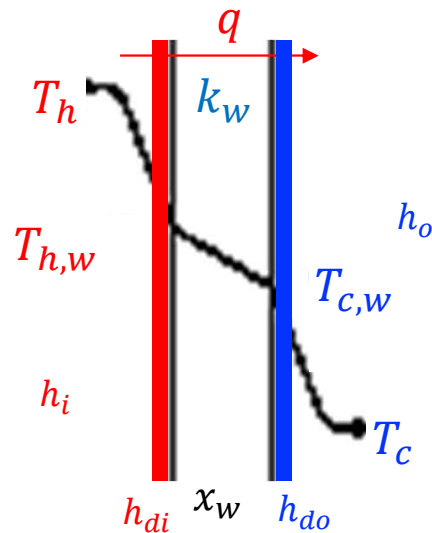
4. Relation between overall heat transfer coefficient and individual heat transfer coefficient

2) with scale deposit

<Assumptions>

1) $dA_i \cong dA_{di}$

2) $dA_o \cong dA_{do}$



$$\delta q = \frac{\Delta T}{R_{total}}$$

$$\Delta T = T_h - T_c$$

$$R_{total} = R_i + R_{di} + R_w + R_{do} + R_o$$

$$R_i = \frac{1}{h_i d A_i} \quad R_{di} = \frac{1}{h_{di} d A_i}$$

$$R_o = \frac{1}{h_o d A_o} \quad R_{do} = \frac{1}{h_{do} d A_o}$$

$$R_w = \frac{x_w}{k_w d A_L}$$

$$\delta q = \frac{\Delta T}{\frac{1}{h_i d A_i} + \frac{1}{h_{di} d A_i} + \frac{x_w}{k_w d A_L} + \frac{1}{h_{do} d A_o} + \frac{1}{h_o d A_o}}$$

$$\frac{1}{U_i} = \frac{1}{h_i} + \frac{1}{h_{di}} + \frac{x_w D_i}{k_w D_L} + \frac{1}{h_o} \frac{D_i}{D_o} + \frac{1}{h_{do}} \frac{D_i}{D_o}$$

$$\frac{1}{U_o} = \frac{1}{h_i} \frac{D_o}{D_i} + \frac{1}{h_{di}} \frac{D_o}{D_i} + \frac{x_w D_o}{k_w D_L} + \frac{1}{h_{do}} + \frac{1}{h_o}$$