

## Effect of SDC Modification on Cathode for Low Temperature Solid Oxide Fuel Cell

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To increase solid oxide fuel cell performance operating at intermediate temperature below 700°C, cathode modifications were performed in electrode microstructure control with an electronic conductor,  $\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$  (LSM), and a mixed conductor,  $\text{La}_{0.6}\text{Sr}_{0.4}\text{Co}_{0.2}\text{Fe}_{0.8}\text{O}_3$  (LSCF). For both cathode materials,  $\text{Sm}_{0.2}\text{Ce}_{0.8}\text{O}_2$  (SDC) buffer layer as a diffusion barrier layer was coated on the yttria-stabilized zirconia (YSZ) electrolyte to prevent the interlayer formation of  $\text{SrZrO}_3$  and  $\text{La}_2\text{Zr}_2\text{O}_7$  which have a poor ionic conductivity. The interfacial reaction products were hardly formed in electrolyte/cathode interlayer by applying SDC layer sintering at high temperature and the cathode polarization was also decreased. Moreover, to enlarge the triple phase boundary, SDC was coated in sol-gel method after sintering the cathode to improve cell performance at low temperature. Cathode resistance for LSCF cathode cell with SDC modification was as low as  $0.11\Omega\text{cm}^2$  in air atmosphere, measured at 700°C. The maximum power densities by cell modifications were  $369\text{mW}/\text{cm}^2$  for LSCF cathode cell, and  $271\text{mW}/\text{cm}^2$  for LSM cathode cell.