

Photocatalytic water splitting by using $(\text{Sr}_x\text{La}_{1-x})\text{TiO}_3$ -SiC system under visible light irradiation

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Energy consumption covers the depletion of fossil fuels and serious environmental problems accompanying their combustion. Hence, a new form of energy that is clean, renewable, cheap, and a viable alternative to fossil fuels is needed. Hydrogen is considered as an ideal fuel for the future. Photocatalytic water splitting is a promising technique to produce hydrogen from water by using photocatalyst.

Silicon carbide (SiC) is not a common photocatalyst for water splitting, eventhough it shows photocatalytic activity which was reported in previous researches. In this research, $(\text{Sr}_x\text{La}_{1-x})\text{TiO}_3$ will be coupled with SiC to increase the photocatalytic activity in splitting the water. Nanosize particles of SiC were mixed with precursors of $(\text{Sr}_x\text{La}_{1-x})\text{TiO}_3$, then spray pyrolyzed at 900°C with 4L/min air flow as gas carrier. SiC is expected to slowing down the electron hole recombination. The photocatalytic activity measurement was conducted under visible light irradiation ($>400\text{ nm}$). XRD, Raman and SEM were conducted to reveal the morphology and elements contained by photocatalyst particles.