## K-CHA type의 zeloite 합성 및 메탄을 thermal cracking에 의한 수소생산

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Production of hydrogen is significant industrial importance, and increasing interest to take advantage of environmentally friendly energy. Particularly, methanol as a hydrogen carrier has been attractively received because its advantages compared to hydrocarbons and ethanol, such as high hydrogen—to—carbon ratio, molecular simplicity, no C—C bond, relatively low reforming temperatures due to its low energy chemical bond, and low sulfur content. Hydrogen production by catalytic methanol thermal cracking has been successfully demonstrated and a large variety of catalysts have been reported. The majority of these have focused on the transition and noble metals—based catalysts. However, there is a serious problem in the cost. Researches used only main metals as a catalyst, the cheaper and harmless to bio—body, are very seldom for the methanol to hydrocarbon reaction. In present study, we have tried to use main metals, such as K, Al, and Si, to synthesize a catalyst with chabasite structure (CHA), and to apply the CHA typed catalyst into methanol to hydrogen reaction. It was concluded that the CHA structured KwAlxSiyOz catalysts gave the better influence to the methanol conversion to hydrogen.