

Prediction of physicochemical properties for jatropha biodiesel + ethan-1-ol or propan-1-ol binary mixtures at T= (288.15 -308.15K) and atmospheric pressure

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The thermophysical properties such as density ( $\rho$ ) and speed of sound ( $v$ ) were measured for binary mixtures (blends) of {jatropha curcas biodiesel (1) + ethan-1-ol or propan-1-ol (2)} over whole composition range at T= (288.15-308.15 K) and atmospheric pressure, using an Anton Paar digital vibrating glass tube densimeter (model DSA 5000). The observed data has been utilized to evaluate the excess molar volume  $V_{12}^E$ , isentropic compressibility  $(\kappa_S)_{12}$  and deviation in isentropic compressibilities  $(\Delta\kappa_S)_{12}$ .

These properties were fitted to Redlich-Kister polynomial equation to calculate the adjustable parameters together with standard deviation. Further, the results have been interpreted in terms of molecular interactions between biodiesel and alcohol molecules.