

Mesoporous MgO-TiO₂ Adsorbents for CO₂ CaptureHiremath Vishwanath, 서정길[†]

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MgO has been extensively reported as an efficient solid CO₂ sorbent. However, it is still challenge to make MgO stable under TSA (Thermal-Swing Adsorption) operation because high temperature is required for regeneration of MgCO₃ to MgO. In this work, TiO₂ stabilized MgO sorbents were investigated as candidate for CO₂ capture. Compositional modifications are carried out to enhance the CO₂ sorption performance of MgO-TiO₂ sorbents. It was observed that the phase was transformed in the sequence of MgTiO₃ (Mg/Ti=0.5-2.0) → MgO-MgTi₂O₄ (Mg/Ti=3.0-4.0) → MgO-Mg₂TiO₄ (Mg/Ti=5.0), and the CO₂ adsorption increased in the order MgTiO₃ < MgO-Mg₂TiO₄ < MgO-MgTi₂O₄. High surface area (111.4 m²/g), large pore volume (0.45 cm³/g), and appropriate basic sites of MgO-MgTi₂O₄ were beneficial for CO₂ adsorption. This work was supported by the National Research Foundation of Korea (NRF) funded by the Ministry of Education (Grant number: NRF-2013R1A1A2060638).