

A novel Ni-doped $\text{Sr}_{0.92}\text{Y}_{0.08}\text{TiO}_3$ catalysts for dry reforming of methane

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Dry (CO₂) reforming of methane (DMR) is an important role in both scientific and industrial field. The syngas from DMR can be used a wide range of industry such as fuel as fuel cell and products of higher alkanes and oxygenates by means Fischer-Tropsch synthesis. The most widely used catalysts for DMR are based on Ni. In general DMR methane with CO₂ is converted into H₂ and CO on the surface of the conventional Ni-based catalyst through a pathway that produces carbon fibers. The factor limiting the commercialization of DMR is poor long-term stability of the catalyst due to the catalyst deactivation because of the formation of carbon deposits on the catalyst surface. Therefore, a new catalyst material is required for DMR having a good long-term stability. A class of materials called the perovskite (ABO₃) have been used to study DMR. The perovskite-based catalyst exhibits excellent coking resistance for DMR due to inherent oxygen mobility. This study demonstrates the development of SYT (Y-substituted SrTiO₃)-based perovskite oxide as advanced catalyst for DMR as function of GHSV (Gas Hourly Space Velocity), temperature, and ratios of CH₄ to CO₂.