Effect of cluster in indium-based metal-organic frameworks for the cycloaddition of epoxides and ${\rm CO}_2$

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 ${\rm CO}_2$ is an advantageous C1 feedstock as it is abundant, inexpensive and common industrial waste. One useful method to add value to ${\rm CO}_2$ is by its ring-opening with epoxides to synthesize cyclic carbonates. Cycloaddition is attractive in terms of ${\rm CO}_2$ utilization as it is catalytic, yields valuable products already used and sold, and results in significant carbon dioxide uptake. Metal-organic frameworks (MOFs) are versatile material formed from organic linker which connect metal. MOFs are prospective candidates in ${\rm CO}_2$ conversion catalyst owing to their rich Lewis acid/base sited and functional group. We report the catalytic efficiency of indium based MOF, denoted as CPM-200. The catalysts was characterized using a various physicochemical analysis, including XRD, XPS, ICP-OES, EA, FE-SEM, TGA, FT-IR, and BET. The catalytic activity of CPM-200 was analyzed in the synthesis of cyclic carbonate from epoxide and ${\rm CO}_2$, and it was shown that CPM-200 operated in synergy effect with co-catalyst tetrabutylammonium bromide (TBAB) under solvent-free conditions. Several reaction parameter studies were carried out to find the optimal condition. Finally, a plausible reaction mechanism was suggested.