Adsorption Studies of 2-Picoline from Taxus chinensis by Sylopute

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Batch experiment studies were carried out on the adsorption of the major tar compound, 2-picoline, derived from the plant cell cultures of Taxus chinensis, using Sylopute while varying parameters such as initial concentration, contact time and adsorption temperature. The experimental data were fitted to the Langmuir, Freundlich, Temkin and Dubinin-Radushkevich isotherm models. The kinetic data were then fitted using the pseudo-first-order, pseudo-second-order and intraparticle diffusion models. Thermodynamic parameters, such as activation energy (Ea), standard enthalpy ($\triangle H^\circ$), standard entropy ($\triangle S^\circ$) and standard Gibbs free energy ($\triangle G^\circ$) change, were also investigated.

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