TiO2-based nanostructured photocatalysts for high photocatalytic degradation of organic dyes

Nanostructured TiO<sub>2</sub> would have large surface area with high pore volumes, which is adequate for adsorbing plenty amount of organic dyes (or pollutants) in water and air. Also, TiO<sub>2</sub> is well-known as a highly reactive photocatalyst under UV light irradiation. For faster photocatalysis process, TiO<sub>2</sub> should be improved its light-harvesting ability, and charge transfer ability to generate reactive oxygen species. Here, other metal oxide (WO<sub>x</sub> and FeO<sub>x</sub>)-incorporated mesoporous TiO<sub>2</sub> were prepared via hard-template method using (1) titanium glycolate spheres and (2) KIT-6 (Ia3d mesostructure). (1) WO<sub>x</sub>-incorporated mesoporous TiO<sub>2</sub> showed faster photocatalytic degradation rate than mesoporous TiO<sub>2</sub> exhibited. This phenomenon would be explained its larger surface area, and higher crystallinity of anatase phase. (2) Ordered mesoporous Fe<sub>2</sub>O<sub>3</sub>-TiO<sub>2</sub> nanocomposites decomposed organic dyes effectively under both UV and visible lights. The heterojunctions between Fe<sub>2</sub>O<sub>3</sub> and TiO<sub>2</sub> nanoparticles would help excitons easily separated into the conduction and valence band, thus, reactive oxygen species could be generated without difficulty.