

## Fabrication of metal oxide loaded CHA catalyst for NH<sub>3</sub>-SCR with LNT function and high thermal stability

박순희, 정하나<sup>1</sup>, 조성준<sup>1,†</sup>

고려대학교; <sup>1</sup>전남대학교

(sjcho@jnu.ac.kr<sup>†</sup>)

Selective catalytic reduction (SCR) with ammonia to remove NO<sub>x</sub> using Cu-exchanged CHA catalyst has been mainly used for diesel vehicles. However, the Cu/CHA catalyst has very poor activity at low temperature, making it more difficult to remove NO<sub>x</sub> by introducing exhaust gas recirculation (EGR) technology, which results in lower exhaust gas temperatures. In addition, since NH<sub>3</sub>-SCR system using liquid urea works well above 200 °C, it is urgent to develop a technology to remove NO<sub>x</sub> at low temperature. Therefore, we fabricated SCR catalysts with LNT function to adsorb and store NO<sub>x</sub> below 200 °C and desorb above 300 °C by loading various metal oxides on CHA. In this study, NO and NH<sub>3</sub> adsorption-desorption behavior of various metal oxides loaded CHA catalyst such as CeO<sub>2</sub>, CaO, MgO, La<sub>2</sub>O<sub>3</sub>, and BaO was investigated by temperature programmed desorption method. In particular, CeO<sub>2</sub> or La<sub>2</sub>O<sub>3</sub> loaded CHA catalysts adsorbed a considerable amount of NO and NH<sub>3</sub> and were very stable enough to maintain the adsorbed amount even after the hydrothermal aging at 750 °C. We also investigated the NH<sub>3</sub>-SCR activity and thermal stability of the catalyst on which copper was additionally loaded.