

**Multi-layered core-shell catalyst for CO<sub>2</sub> reforming of methane**

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The catalytic CO<sub>2</sub> reforming of methane possesses an eco-friendly implication to produce syngas (H<sub>2</sub> and CO) by converting both major greenhouse gases CH<sub>4</sub> and CO<sub>2</sub>. However, the catalyst deactivation by sintering and carbon coking is the major issue to implement scale-up process in real applications. In this study, we prepared one-pot synthesized catalysts consisting of single core@shell and multi-layered catalysts sequentially synthesized (dual and triple-layered core@shell), while keeping a total loading of Ni (*ca.* 10 wt.%) in the sample. The Ni impregnated on SiO<sub>2</sub> (Ni/SiO<sub>2</sub>) catalyst was also compared. The CO<sub>2</sub> reforming of methane reaction was carried out under the steady-state condition from 650 to 850 °C with a mixture of CH<sub>4</sub> and CO<sub>2</sub> in Ar balance (CH<sub>4</sub>:CO<sub>2</sub>:Ar=3:3:4). The catalytic performance is in the following order: Dual layered Ni@SiO<sub>2</sub> > triple-layered Ni@SiO<sub>2</sub> > single Ni@SiO<sub>2</sub> > Ni/SiO<sub>2</sub>. Moreover, the dual-layered core-shell catalyst revealed superior long-term stability under the reaction condition with SV=100,000 h<sup>-1</sup> for 100 h time-on-stream. The catalysts were also characterized by H<sub>2</sub>-TPR, NH<sub>3</sub>-TPD, XPS, and TEM to analyze the activity trend and the catalytic stability.